## LIGAND PROPERTIES OF THE -OSeF<sub>5</sub> AND -OTeF<sub>5</sub> GROUPS IN PSEUDO-TRIGONAL-BIPYRAMIDAL MOLECULES

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The compounds  $F_2\text{Te}(0\text{Te}F_5)$  and  $F_2\text{Se}(0\text{Te}F_5)_2$  are generated by ligands exchange starting with  $\text{Te}F_4$  (SeF $_4$ ) and B(0TeF $_5$ ) $_3$ .

 ${
m F_2Se(0SeF_5)_2}$  has been prepared by reactions of  ${
m Hg(0SeF_5)_2}$  and  ${
m SeCl}_4$ 

$$SeC1_4 + 2Hg(OSeF_5)_2 \xrightarrow{-2HgC1_2} \left\{ Se(OSeF_5)_4 \right\}$$

$$F_4Se \xrightarrow{0} SeF_4 + F_2Se(OSeF_5)_2$$

All three compounds are structurally related to  $SeF_4$  and  $TeF_4$ . There is a pair of axial and a pair of equatorial ligands and a no-bonding electron pair.

The structures have been investigated by  $^{77}$ Se and  $^{125}$ Te n.m.r. spectroscopy. All three materials have the expected pseudo-trigonal-bipyramidal structures with  $\underline{\text{axial}}$  -0SeF $_5$  (-0TeF $_5$ ) ligands, and the fluorine ligands in  $\underline{\text{equatorial}}$  positions.

This behaviour is discussed in terms of the extreme electronegativity of these ligands.